



Common Conversions and Constants

Air

$C_p = 0.2445 \text{ BTU} / \text{lb} - ^\circ\text{F}$
Density = 0.075 Lb / CF
Spec. Vol. = 13.3 CF / Lb

Water

$C_p = 1.0 \text{ BTU} / \text{lb} - ^\circ\text{F}$

Fuel Heat Contents

1.03 MMBTU / MCF (Natural Gas)
0.135 MMBTU / Gallon (#2 Oil or "Diesel")
0.092 MMBTU / Gallon (Propane)
2.480 MMBTU / MCF (Propane)
3.20 MMBTU / MCF (Butane)

Steam

Heat Of Vaporization
Atmospheric Pressure Steam 970 BTU / lb
15 Psig Steam 945 BTU / lb
100 Psig Steam 880 BTU / lb
1 Boiler HP = 33,475 BTU / Hr = 10 kW = 34 lbs of steam / hr
at 212°F

Sensible Heat

The sensible heat in a heating or cooling process of air (heating or cooling capacity) can be expressed as

$$h_s = 1.08 q dt \quad (1)$$

where

$$h_s = \text{sensible heat (Btu/hr)}$$

$$q = \text{air volume flow (cfm, cubic feet per minute)}$$

$$dt = \text{temperature difference (}^\circ\text{F)}$$

Total Heat - Latent and Sensible Heat

Total heat due to both temperature and moisture can be expressed as:

$$h_t = 4.5 q dh \quad (4)$$

where

$$h_t = \text{total heat (Btu/hr)}$$

$$q = \text{air volume flow (cfm, cubic feet per minute)}$$

$$dh = \text{enthalpy difference (btu/lb dry air)}$$

Total heat can also be expressed as:

$$h_t = h_s + h_l \\ = 1.08 q dt + 0.68 q dw_{gr} \quad (5)$$

Volume and Weights

7.48 gal / CF
8.34 lb / gal H₂O
1.0 MCF = 1000 CF
1.0 lb = 7000 grains

Pressure

psig = 2.31 ft head (water) = 27.72 in. W.G.
in. HG = 13.61 in. W.G.

Time

8,760 hrs / year
52 weeks / year
730 hrs / month

Energy/ Power

12,000 BTU / ton-hr
0.746 KW / HP
3,413 BTU / KWH

Distance and Area

Acre = 43,560 ft²
mile = 5,280 feet

Latent Heat

The latent heat due to moisture in the air can be expressed as:

$$h_l = 0.68 q dw_{gr} \quad (2)$$

or

$$h_l = 4,840 q dw_{lb} \quad (3)$$

where

$$h_l = \text{latent heat (Btu/hr)}$$

$$q = \text{air volume flow (cfm, cubic feet per minute)}$$

$$dw_{gr} = \text{humidity ratio difference (grains water/lb dry air)}$$

$$dw_{lb} = \text{humidity ratio difference (lb water/lb dry air)}$$

unds • 1 grain = 0.000143 lb = 0.0648 g

$$1 \text{ hp} = .746 \text{ KW} = \text{KW} \div 1.341$$

$$1 \text{ hp} = 2,547 \text{ BTU per hour}$$

$$1 \text{ BTU} = \text{Heat required to raise 1 lb water } 1 \text{ }^\circ\text{F}$$

$$1 \text{ BTU} = 777.6 / \text{Foot-pounds work}$$

$$1 \text{ Kilowatt Hour} = 3,415 \text{ BTU}$$

$$\text{Heat Value of Carbon} = 14,600 \text{ BTU per pound}$$

$$\text{Latent Heat of Fusion of Ice} = 143.15 \text{ BTU per pound}$$

$$\text{Latent Heat of Evaporation of Water at } 212 \text{ }^\circ\text{F.} = 970.4 \text{ BTU per pound}$$

$$\text{Total Heat of Saturated Steam at Atmospheric Pressure} = 1,150.4 \text{ BTU per pound}$$

$$1 \text{ Ton of Refrigeration} = 288,000 \text{ BTU per 24 hours}$$

$$g = \text{Acceleration of Gravity (commonly taken as } 32.16 \text{ feet per second per second)}$$

$$1 \text{ Radian} = 57.296^\circ$$

$$1 \text{ Meter} = 100 \text{ cm} = 39.37 \text{ in}$$

$$1 \text{ Kilometer} = .62137 \text{ mi}$$

$$1 \text{ Gallon} = 231 \text{ cu in}$$

$$1 \text{ Barrel} = 31.5 \text{ gal}$$

$$\text{Atmospheric Pressure} = 14.7 \text{ lbs per sq in} = 29.92 \text{ in mercury at } 32 \text{ }^\circ\text{F}$$

$$1 \text{ lb per sq in Pressure} = 2.3095 \text{ ft fresh water at } 62 \text{ }^\circ\text{F}$$

$$= 2.0355 \text{ in mercury at } 32 \text{ }^\circ\text{F}$$

$$= 2.0416 \text{ in mercury at } 62 \text{ }^\circ\text{F}$$

$$\text{Water Pressure (lbs per sq in)} = .433 \times \text{height of water in ft (Fresh water at } 62 \text{ }^\circ\text{F)}$$

$$\text{Weight of 1 cu ft Fresh Water} = 62.355 \text{ lbs at } 62 \text{ }^\circ\text{F} = 59.76 \text{ lbs at } 212 \text{ }^\circ\text{F}$$

$$\text{Weight of 1 cu ft Air at } 14.7 \text{ lbs per sq in Pressure} = .07608 \text{ lbs. at } 62 \text{ }^\circ\text{F} = .08073 \text{ lbs at } 32 \text{ }^\circ\text{F}$$